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Asymmetric ruthenium-catalysed [2+2] cycloadditions between bicyclic alkenes and chiral aryl-substituted acetylenic acyl camphorsultam alkynes

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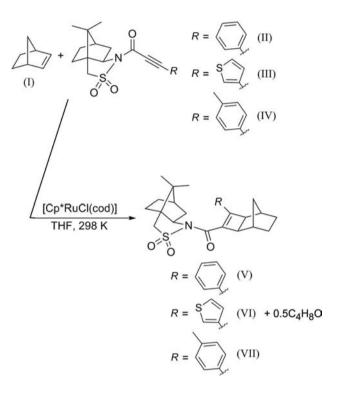
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The regio- and absolute stereochemistry of (7S)-N-[4-(3thienyl)tricyclo[4.2.1.0^{2,5}]non-3-en-3-ylcarbonyl]-2,10-camphorsultam tetrahydrofuran hemisolvate, C₂₄H₂₉NO₃S₂·0.5C₄H₈O, and (7S)-N-[4-(4-tolyl)tricyclo[4.2.1.0^{2,5}]non-3-en-3-ylcarbonyl]-2,10-camphorsultam, C₂₇H₃₃NO₃S, have been established. One contains a half-occupancy tetrahydrofuran solvent molecule located on a twofold axis and the other contains two crystallographically unique molecules which are nearly identical. The extended structures of both complexes can be explained via weak C-H···O interactions, which link the molecules together into two-dimensional sheets in the ab plane for the thienvl complex and ultimately into a threedimensional structure for the tolvl derivative. The stereochemistry of both structures confirms that [2+2] cycloadditions of bicyclic alkenes and alkynes catalysed by ruthenium are exclusively exo.

Comment

Recently, the metal-catalysed [2+2] cycloaddition reaction between bicyclic alkenes and alkynes has been utilized by several research groups as a means to generate cyclobutene rings. In terms of an asymmetric version of this process, both a chiral rhodium catalyst (Shibata *et al.*, 2006) and a rutheniumcatalysed reaction (Villeneuve & Tam, 2004) involving a chiral precursor have produced chiral [2+2] cyclobutene products. The ruthenium-catalysed [2+2] cycloaddition between norbornene and a phenylacetyleneacyl camphorsultam (chiral) yielded two diastereomers in a 131:1 ratio. The absolute stereochemistry of the major diastereomer (V) was established by single-crystal X-ray diffraction (Lough *et al.*, 2004).

As an extension of this study, the Tam group is investigating the selectivity effect of different organic groups attached at the acetylenic position (see scheme). In this paper, we present the results of ruthenium-catalysed [2+2] cycloadditions of norbornene, (I), with 3-thiophenylacetyleneacylcamphorsultam, (III), and 4-tolylacetyleneacylcamphorsultam, (IV). This yielded the expected complexes (7S)-N-[4-(3-thienyl)tricyclo[4.2.1.0^{2,5}]non-3-en-3-ylcarbonyl]-2,10-camphorsultamtetrahydrofuran hemisolvate, (VI), and <math>(7S)-N-[4-(4-tolyl)tricyclo[4.2.1.0^{2,5}]non-3-en-3-ylcarbonyl]-2,10-camphorsultam,(VII). Two diastereomers of (VI) were obtained in a 20:1 ratio,



whereas the major stereoisomer of (VII) was obtained in a 104:1 ratio over the minor isomer. The absolute stereochemistry of the major isomers of (VI) and (VII) was estab-

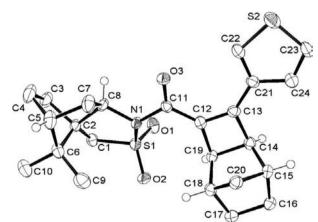


Figure 1

A view of the structure of (VI), showing the crystallographic labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. For the sake of clarity, nonchiral H atoms have been omitted, as has the partial-occupancy THF molecule of solvation.

lished by the single-crystal X-ray diffraction analysis described here.

A view of (VI) is shown in Fig. 1. Elucidation of the X-ray structure establishes the absolute configuration of the following atoms in Fig. 1: C2-S, C5-R, C8-R, C14-S, C15-R, C18-S and C19-R. This is the same configuration as that observed for the phenyl derivative (V) (Lough *et al.*, 2004). There is a tetrahydrofuran (THF) solvent molecule occluded in the structure. It was modelled at half-occupancy, as it is located about $(\frac{1}{2}, \frac{1}{2}, 0)$, which lies on a twofold axis along the [110] vector. The THF molecule was refined giving reasonable displacement parameters; the O atom of this molecule is involved in the weak interactions discussed below.

There are two chemically identical molecules of (VII) in the asymmetric unit. They are very similar structurally, with an r.m.s. fit of 0.207 Å (*PLATON*; Spek, 2009). As expected, both have the same configuration as that observed for (VI). The absolute configuration, as determined by anomalous dispersion, is as follows: C2-*S*, C5-*R*, C8-*R*, C14-*S*, C15-*R*, C18-*S* and C19-*R* for one molecule, and C32-*S*, C35-*R*, C38-*R*, C44-*S*, C45-*R*, C48-*S* and C49-*R* for the second. The structure of one of the molecules of (VII) is shown in Fig. 2. The crystallographic difference in the two molecules is primarily attributed to their spatially different orientations with respect to one another (see below). The geometric data for both (VI) and (VII) are in the normal ranges.

Weak C-H···O interactions in (VI) (see Table 1 for geometric parameters and symmetry codes) link the molecules. Perhaps the simplest method to describe the packing is by using two 'classical model' graph-set descriptors (Bernstein *et al.*, 1995). Firstly, a 'chain of rings' is formed through two weak interactions (C8-H8A···O2ⁱ and C1-H1B···O3ⁱⁱ) that link molecules together into ribbons along the *a* axis {motif $C_2^2(10)[R_2^2(9)]$; Fig. 3}. Secondly, adjacent ribbons are linked together *via* reciprocating C14-H14A···O1ⁱⁱⁱ interactions. It should be noted that these are the 'weakest' of the weak interactions mentioned herein. There are two further layers in

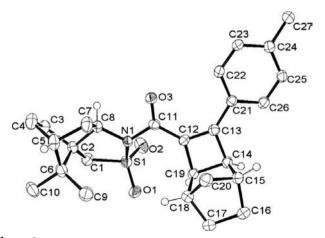


Figure 2

The molecular structure of one of the molecules of (VII), showing the atom-labelling scheme and with displacement ellipsoids drawn at the 50% probability level. For the sake of clarity, the nonchiral H atoms have been omitted.

the extension along the *c* axis, each of which is rotated by 120° with respect to the previous layer. The THF solvent molecule was refined at half-occupancy, as it is located on a rotational axis; the O atom links to one of the layers *via* another weak interaction (C1-H1A···O31).

As with (VI), it is the weak $C-H\cdots O$ interactions that control the geometry observed in the crystal structure of (VII) (Table 2). The nine weak interactions lead to a very extensive set of hydrogen-bonding motifs so only the key graph-set descriptors are noted here. The asymmetric unit contains two molecules, which couple together *via* two weak interactions $(C3-H3B\cdots O31$ and $C31-H31B\cdots O3$) to form an 11membered ring, and, adjacent to this, a nine-membered ring is formed *via* $C38-H38A\cdots O1^{i}$ and $C1-H1B\cdots O33^{ii}$ interactions. This quaternary graph set, somewhat similar to that in (VI), has the motifs $C_2^2(10)[R_2^2(11)R_2^2(9)]$: a 'chain of rings' yielding ribbons along the *a* axis (Fig. 4). These *a*-axis ribbons are connected up and down the *c* axis according to the binary

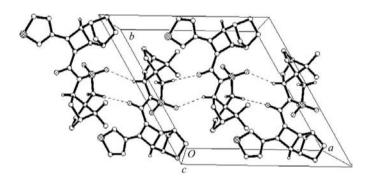


Figure 3

The packing of (VI), showing the primary chain-forming $C-H \cdots O$ interactions as dashed lines. All H atoms, except those attached to atoms involved in weak interactions, have been omitted for clarity. The view is approximately down the *c* axis, showing the ribbons along the *a* axis.

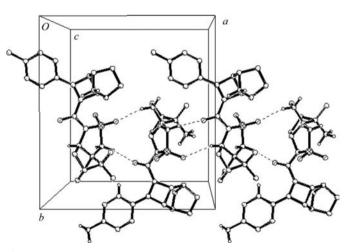


Figure 4

The packing of (VII), showing the primary chain-forming $C-H\cdots O$ interactions as dashed lines. All H atoms, except those involved in the displayed weak interactions, have been omitted for clarity. The view is approximately down the *c* axis, showing the formation of a chain along the *a* axis.

graph-set descriptor $C_2^2(12)$ via C50-H50A···O2ⁱⁱⁱ and C9- $H9A \cdots O3^{iv}$ interactions, thus forming parallel ribbons in the [010] plane. Finally, a full three-dimensional network is established via C27-H27A···O32^v and C9-H9A···O3^{iv} interactions, making $C_2^2(25)$ connections along the b axis. For completeness, the last two interactions in Table 2 can be described by motif S(7).

While there are no classical hydrogen-bonding networks in either of these structures, the weak interactions do seem to explain the observed geometry, which is understandable as even a weak hydrogen bond will be stronger than the van der Waals interactions. Unlike the previously published phenyl cycloadduct which shows partial $\pi - \pi$ stacking between neighbouring phenyl rings (closest contact = 3.64 Å) through secondary orbital overlap interactions (Lough et al., 2004), no such π -stacking interactions between either thiophenyl rings or tolyl rings are observed in these crystal structures.

In the case of the [2+2] cycloaddition, these crystal structures have unambiguously identified which are the major diastereomers formed in the asymmetric cycloaddition. The previous phenyl derivative combined with both the thiophenyl and the tolyl derivatives herein reaffirm that [2+2] cycloadditions of bicyclic alkenes and alkynes catalysed by ruthenium are exclusively exo with respect to their stereochemistry.

Experimental

Preparation of compound (VI) was achieved by addition of norbornene (72.0 mg, 765 mmol), (I), and chiral alkyne (III) (52.2 mg, 0.149 mmol) to Cp*Ru(cod)Cl (Cp* is pentamethylcyclopentadienyl and cod is 1,5-cyclooctadiene; 7.6 mg, 0.020 mmol) in THF (0.6 ml). The mixture was stirred for 18 h at 298 K, and subsequent column chromatography of the crude reaction mixture provided a 98% yield of the two diastereomeric cycloadducts in a 20:1 ratio. Evaporative recrystallization of the column-purified product in THF provided the major diastereomer (VI), giving colourless crystals suitable for X-ray analysis. Similarly, compound (VII) was prepared by addition of norbornene (83.0 mg, 0.881 mmol), (I), and chiral alkyne (IV) (63.2 mg, 0.177 mmol) to Cp*Ru(cod)Cl (7.1 mg, 0.018 mmol) in THF (0.5 ml). The mixture was stirred for 17 h at 298 K, and subsequent column chromatography of the crude reaction mixture provided a 79% yield of the two diastereomeric cycloadducts in a 104:1 ratio. Evaporative recrystallization of the column-purified product in THF provided major diastereomer (VII), giving colourless crystals suitable for X-ray analysis. Both diastereomeric ratios were back-calculated from the corresponding ee values obtained through cleavage of the chiral auxiliary under reductive conditions (LiAlH₄, AlCl₃), giving an enantiomeric mixture of cycloadduct alcohols that can be resolved using a Chiralcel OJ-H high-performance liquid chromatography column.

Compound (VI)

Crystal data

C24H29NO3S2.0.5C4H8O $M_r = 479.65$ Trigonal, P3₂21 a = 12.0797 (17) Åc = 28.326 (6) Å $V = 3579.5 (10) \text{ Å}^3$

Z = 6Mo $K\alpha$ radiation $\mu = 0.25 \text{ mm}^ T=150~{\rm K}$ $0.35 \times 0.22 \times 0.15 \text{ mm}$

Table 1

Hydrogen-bond geometry (Å, °) for (VI).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$C8-H8A\cdots O2^{i}$	1.00	2.59	3.516 (3)	154
$C1 - H1B \cdots O3^{ii}$	0.99	2.45	3.422 (3)	168
$C14-H14A\cdots O1^{iii}$	1.00	2.66	3.500 (3)	141
$C1-H1A\cdots O31$	0.99	2.36	3.235 (14)	147

Symmetry codes: (i) $x - y, -y + 1, -z + \frac{1}{3}$; (ii) $x - y + 1, -y + 1, -z + \frac{1}{3}$; (iii) x - y + 1, $-v + 2, -z + \frac{1}{2}$

Table 2

Hydrogen-bond geometry (Å, °) for (VII).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C3-H3B\cdots O31$	0.99	2.58	3.539 (3)	163
C31-H31B···O3	0.99	2.45	3.356 (2)	152
$C38-H38A\cdotsO1^{i}$	1.00	2.54	3.412 (2)	146
$C1 - H1B \cdot \cdot \cdot O33^{ii}$	0.99	2.37	3.332 (3)	164
$C50-H50A\cdots O2^{iii}$	0.99	2.53	3.446 (2)	153
C9−H9A···O3 ^{iv}	0.98	2.59	3.570 (3)	174
$C27 - H27A \cdots O32^{v}$	0.98	2.43	3.404 (3)	174
C22-H22A···O3	0.95	2.31	3.115 (3)	142
C52-H52A···O33	0.95	2.33	3.129 (3)	141

Symmetry codes: (i) x - 1, y, z; (ii) x + 1, y, z; (iii) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$; (iv) $x + \frac{1}{2}$, $-y + \frac{3}{2}, -z + 2;$ (v) x, y + 1, z.

Data collection

Nonius KappaCCD diffractometer	8354 measured
Absorption correction: multi-scan	4207 independ
(DENZO-SMN; Otwinowski &	3310 reflection
Minor, 1997)	$R_{\rm int} = 0.041$
$T_{\min} = 0.916, \ T_{\max} = 0.963$	

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$	H-atom parameters constrained
$wR(F^2) = 0.084$	H-atom parameters constrained $\Delta \rho_{\text{max}} = 0.19 \text{ e} \text{ Å}^{-3}$
S = 1.01	$\Delta \rho_{\rm min} = -0.31 \text{ e } \text{\AA}^{-3}$
4207 reflections	Absolute structure: Flack (1983),
318 parameters	1788 Friedel pairs
4 restraints	Flack parameter: -0.03 (7)

Compound (VII)

Crystal data

C₂₇H₃₃NO₃S V = 4737.2 (16) Å³ $M_r = 451.60$ Z = 8Orthorhombic, $P2_12_12_1$ a = 11.687 (2) Å b = 12.849 (3) Å c = 31.547 (6) Å

Data collection

Nonius KappaCCD diffractometer Absorption correction: multi-scan (DENZO-SMN; Otwinowski & Minor, 1997) $T_{\min} = 0.937, \ T_{\max} = 0.972$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.041$ $wR(F^2) = 0.096$ S = 1.039673 reflections 583 parameters H-atom parameters constrained d reflections dent reflections ns with $I > 2\sigma(I)$

Mo $K\alpha$ radiation $\mu = 0.17 \text{ mm}^{-1}$ T = 150 K $0.40 \times 0.35 \times 0.17 \text{ mm}$

44242 measured reflections 9673 independent reflections 7676 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.068$

 $\Delta \rho_{\text{max}} = 0.19 \text{ e} \text{ Å}^{-3}$ $\Delta \rho_{\rm min} = -0.31$ e Å⁻³ Absolute structure: Flack (1983), 4291 Friedel pairs Flack parameter: -0.02 (5)

Both the O–C bonds and the C–C bonds of the disordered solvent (THF) were restrained, respectively, to be identical. H atoms were positioned geometrically and refined using a riding model, with C–H distances of 0.98 (CH₃), 0.99 (CH₂), 1.00 (CH) or 0.95 Å (aromatic CH). $U_{iso}(H)$ values were set at 1.2 times $U_{eq}(C)$ for all except the methyl H atoms, for which $U_{iso}(H) = 1.5U_{eq}(C)$.

For both compounds, data collection: *COLLECT* (Nonius, 1997); cell refinement: *DENZO–SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO–SMN*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL/PC* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL/PC*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: GA3134). Services for accessing these data are described at the back of the journal.

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